

**LAMPIRAN B**  
**PERHITUNGAN**

**B.1 PethitunganPembuatanlarutan NaOH untuk Proses Delignifikasi**

**a. Untuk proses delignifikasi**

**1. Larutan NaOH 5%**

$$\begin{array}{r} \boxed{2} . \boxed{2} \\ 98\% . \boxed{2} \\ \hline \boxed{2} \end{array} \quad \begin{array}{r} \boxed{2} . \boxed{2} \\ 5\% . 150 \boxed{2} \\ \hline 7,65 \boxed{2} \end{array}$$

**b. Untuk Proses Hidrolisis**

**1.Larutan H<sub>2</sub>SO<sub>4</sub> 2% dalam 120 ml**

$$\begin{array}{r} \boxed{2} . \boxed{2} \\ 98\% . \boxed{2} \\ \hline \boxed{2} \end{array} \quad \begin{array}{r} \boxed{2} . \boxed{2} \\ 2\% . 120 \boxed{2} \\ \hline 2,44 \boxed{2} \end{array}$$

**2. Larutan H<sub>2</sub>SO<sub>4</sub> 3% dalam 120 ml**

$$\begin{array}{r} \boxed{2} . \boxed{2} \\ 98\% . \boxed{2} \\ \hline \boxed{2} \end{array} \quad \begin{array}{r} \boxed{2} . \boxed{2} \\ 3 \% . 120 \boxed{2} \\ \hline 3,67 \boxed{2} \end{array}$$

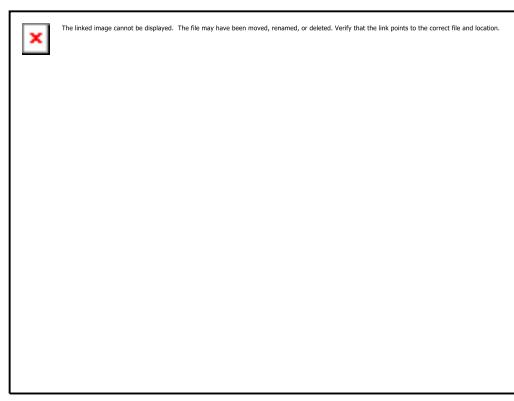
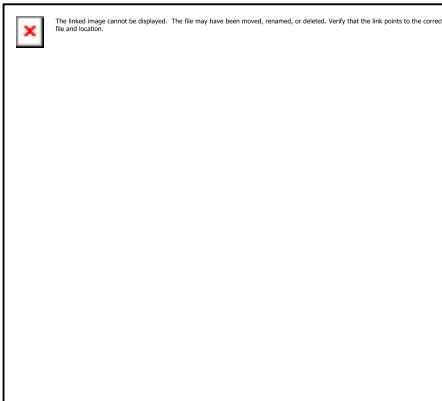
**. 3. Larutan H<sub>2</sub>SO<sub>4</sub> 4% dalam 120 ml**

$$\begin{array}{r} \boxed{2} . \boxed{2} \\ 98\% . \boxed{2} \\ \hline \boxed{2} \end{array} \quad \begin{array}{r} \boxed{2} . \boxed{2} \\ 4 \% . 120 \boxed{2} \\ \hline 4,89 \boxed{2} \end{array}$$

## LAMPIRAN C

### GAMBAR

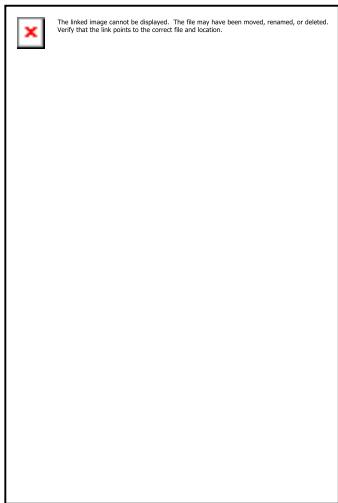
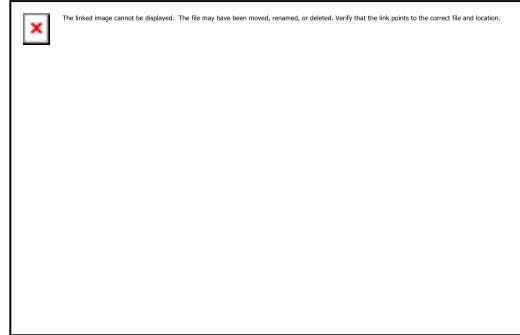
#### C.1 Proses Pembuatan Bioetanol



1.Tandan Kosong Kelapa Sawit di jemur di 2.Proses pengurangan kadar air pada bawah sinarmatahari TKKS dengan cara di oven



3.TKKS yang telah kering dan halus di timbang  
4.Proses Delignifikasi dengan NaOH sebanyak 25 gram

5. Proses Hidrolisisdengan  $H_2SO_4$ 

6. Proses Fermentasi



7. Proses Destilasi

