

## LAMPIRAN A DATA PENGAMATAN

Tabel A.1 Hasil Uji Nilai COD Pada Biogas

No.	Tanggal Sampel	Komposisi Sampel	Parameter Uji	Metode Uji	Vol. FAS Sampel	Vol. FAS Blanko	Vol. Std.	Hasil Uji
1.	23 April 2022				10,1 ml	11 ml	25,1 ml	35856 mg/L
2.	28 April 2022				10,5 ml	11,3 ml	25 ml	32000 mg/L
3.	3 Mei 2022	Kotoran Sapi : Air (1 : 2)	COD ( <i>Chemical Oxygen Demand</i> )	Titrimetri	10,6 ml	11,4 ml	25,2 ml	31744 mg/L
4.	8 Mei 2022				10,7 ml	11,5 ml	25,4 ml	31488 mg/L
5.	13 Mei 2022				11,3 ml	12 ml	24,8 ml	28224 mg/L
6.	18 Mei 2022				11,4 ml	12 ml	24,8 ml	24192 mg/L

Tabel A.2 Hasil Perhitungan Massa Jenis Biogas

No.	Tanggal Sampel	Komposisi Sampel	$\rho$ CH <sub>4</sub> (kg/m <sup>3</sup> )	$\rho$ CO <sub>2</sub> (kg/m <sup>3</sup> )	Hasil Uji (kg/m <sup>3</sup> )
1.	28 April 2022				1,7388054
2.	3 Mei 2022	Kotoran			1,7035854
3.	8 Mei 2022	Sapi : Air	0,668	1,842	1,3246182
4.	13 Mei 2022	(1 : 2)			1,2531216
5.	18 Mei 2022				1,2061616

Tabel A.3 Hasil Pengamatan Tekanan dan pH Pada Biogas

No.	Tanggal Sampel	Komposisi Sampel	Tekanan (bar)	pH
1.	23 April 2022		0	6
2.	28 April 2022		0,1	6
3.	3 Mei 2022	Kotoran Sapi : Air (1 : 2)	0,25	6
4.	8 Mei 2022		0,2	7
5.	13 Mei 2022		0,4	7
6.	18 Mei 2022		0,45	7

Tabel A.4 Hasil Analisa Biogas Menggunakan Alat *Biogas Analyzer*

No.	Tanggal Sampel	Jenis Sampel	Parameter Uji	Metode Uji	Hasil Uji				
1.	28 April 2022		O <sub>2</sub> (%)		20,31				
			CO <sub>2</sub> (%)		12,02				
			CH <sub>4</sub> (%)		8,79				
			H <sub>2</sub> S (ppm)		12,00				
			Net cal. val (MJ/kg)		2,10				
			Gross cal. val (MJ/kg)		2,70				
			Net cal. val (MJ/m <sup>3</sup> )		3,40				
			Gross cal. val (MJ/m <sup>3</sup> )		4,30				
			O <sub>2</sub> (%)		20,18				
			CO <sub>2</sub> (%)		0,04				
2.	3 Mei 2022		CH <sub>4</sub> (%)		11,79				
			H <sub>2</sub> S (ppm)		17,00				
			Net cal. val (MJ/kg)		6,70				
			Gross cal. val (MJ/kg)		7,40				
			Net cal. val (MJ/m <sup>3</sup> )		8,20				
			Gross cal. val (MJ/m <sup>3</sup> )		9,10				
			O <sub>2</sub> (%)		9,31				
			CO <sub>2</sub> (%)		31,09				
			CH <sub>4</sub> (%)		44,07				
			3.		8 Mei 2022	Biogas	H <sub>2</sub> S (ppm)	<i>Biogas Analyzer</i>	2224,00
Net cal. val (MJ/kg)	11,80								
Gross cal. val (MJ/kg)	13,40								
Net cal. val (MJ/m <sup>3</sup> )	13,30								
Gross cal. val (MJ/m <sup>3</sup> )	14,90								
O <sub>2</sub> (%)	11,72								
CO <sub>2</sub> (%)	38,78								
CH <sub>4</sub> (%)	50,16								
H <sub>2</sub> S (ppm)	1805,00								
4.	13 Mei 2022			Net cal. val (MJ/kg)					19,50
			Gross cal. val (MJ/kg)	21,80					
			Net cal. val (MJ/m <sup>3</sup> )	18,10					
			Gross cal. val (MJ/m <sup>3</sup> )	20,00					
			O <sub>2</sub> (%)	8,28					
			CO <sub>2</sub> (%)	37,34					
			CH <sub>4</sub> (%)	54,16					
			H <sub>2</sub> S (ppm)	1034,00					
			5.	18 Mei 2022		Net cal. val (MJ/kg)			18,40
						Gross cal. val (MJ/kg)			20,60
Net cal. val (MJ/m <sup>3</sup> )	19,80								
Gross cal. val (MJ/m <sup>3</sup> )	22,20								

## LAMPIRAN B PERHITUNGAN

### 1. Mencari Nilai COD

#### a. Normalitas Ferro Ammonium Sulfat (FAS)

Diketahui :

Tabel B.1 Hasil Standarisasi Larutan FAS

No.	Tanggal Sampel	Volume $K_2Cr_2O_7$ (ml)	N $K_2Cr_2O_7$ (N)	Volume Titran (ml)
1.	23 April 2022			25,1
2.	28 April 2022			25
3.	3 Mei 2022	10	0,25	25,2
4.	8 Mei 2022			25,4
5.	13 Mei 2022			24,8
6.	18 Mei 2022			24,8

Rumus (SNI, 2004) :

$$N \text{ FAS} = \frac{ml \text{ } K_2Cr_2O_7 \times N \text{ } K_2Cr_2O_7}{ml \text{ standarisasi larutan FAS}}$$

- 23 April 2022

$$N \text{ FAS} = \frac{ml \text{ } K_2Cr_2O_7 \times N \text{ } K_2Cr_2O_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{25,1 \text{ ml}}$$

$$N \text{ FAS} = 0,0996 \text{ N}$$

- 28 April 2022

$$N \text{ FAS} = \frac{ml \text{ } K_2Cr_2O_7 \times N \text{ } K_2Cr_2O_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{25 \text{ ml}}$$

$$N \text{ FAS} = 0,1 \text{ N}$$

- 3 Mei 2022

$$N \text{ FAS} = \frac{ml \text{ K}_2\text{Cr}_2\text{O}_7 \times N \text{ K}_2\text{Cr}_2\text{O}_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{25,2 \text{ ml}}$$

$$N \text{ FAS} = 0,0992 \text{ N}$$

- 8 Mei 2022

$$N \text{ FAS} = \frac{ml \text{ K}_2\text{Cr}_2\text{O}_7 \times N \text{ K}_2\text{Cr}_2\text{O}_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{25,4 \text{ ml}}$$

$$N \text{ FAS} = 0,0984 \text{ N}$$

- 13 Mei 2022

$$N \text{ FAS} = \frac{ml \text{ K}_2\text{Cr}_2\text{O}_7 \times N \text{ K}_2\text{Cr}_2\text{O}_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{24,8 \text{ ml}}$$

$$N \text{ FAS} = 0,1008 \text{ N}$$

- 18 Mei 2022

$$N \text{ FAS} = \frac{ml \text{ K}_2\text{Cr}_2\text{O}_7 \times N \text{ K}_2\text{Cr}_2\text{O}_7}{ml \text{ standarisasi larutan FAS}}$$

$$N \text{ FAS} = \frac{10 \text{ ml} \times 0,25 \text{ N}}{24,8 \text{ ml}}$$

$$N \text{ FAS} = 0,1008 \text{ N}$$

Tabel B.2 Normalitas FAS

No.	Tanggal Sampel	N FAS
1.	23 April 2022	0,0996 N
2.	28 April 2022	0,1 N
3.	3 Mei 2022	0,0992 N
4.	8 Mei 2022	0,0984 N
5.	13 Mei 2022	0,1008 N
6.	18 Mei 2022	0,1008 N

## b. Nilai COD

Diketahui :

Tabel B.3 Hasil Titrasi Sampel dan Blanko

No.	Tanggal Sampel	Volume Titran Blanko (a) (ml)	Volume Titran Sampel (b) (ml)	N FAS (N)	Volume Sampel (ml)
1.	23 April 2022	11	10,1	0,0996	10
2.	28 April 2022	11,3	10,5	0,1	
3.	3 Mei 2022	11,4	10,6	0,0992	
4.	8 Mei 2022	11,5	10,7	0,0984	
5.	13 Mei 2022	12	11,3	0,1008	
6.	18 Mei 2022	12	11,4	0,1008	

Rumus (SNI, 2004) :

$$\text{COD} = \frac{(a-b)(N \text{ FAS})(8000)(P)}{\text{ml sampel}}$$

- 23 April 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(11 - 10,1) \text{ ml} \times 0,0996 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 35856 \text{ mg/L O}_2$$

- 28 April 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(11,3 - 10,5) \text{ ml} \times 0,1 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 32000 \text{ mg/L O}_2$$

- 3 Mei 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(11,4 - 10,6) \text{ ml} \times 0,0992 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 31744 \text{ mg/L O}_2$$

- 8 Mei 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(11,5 - 10,7) \text{ ml} \times 0,0984 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 31488 \text{ mg/L O}_2$$

- 13 Mei 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(12 - 11,3) \text{ ml} \times 0,1008 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 28224 \text{ mg/L O}_2$$

- 18 Mei 2022

$$\text{COD} = \frac{(a-b)(N)(8000)(P)}{\text{ml sampel}}$$

$$\text{COD} = \frac{(12 - 11,4) \text{ ml} \times 0,1008 \text{ N} \times 8000 \times 500 \text{ ml}}{10 \text{ ml}}$$

$$\text{COD} = 24192 \text{ mg/L O}_2$$

Tabel B.4 Nilai COD

No.	Tanggal Sampel	COD (mg/L O <sub>2</sub> )
1.	23 April 2022	35856
2.	28 April 2022	32000
3.	3 Mei 2022	31744
4.	8 Mei 2022	31488
5.	13 Mei 2022	28224
6.	18 Mei 2022	24192

## 2. Mencari Nilai Massa Jenis Biogas

Diketahui :

Tabel B.5 Massa Jenis dan Konsentrasi CH<sub>4</sub> dan CO<sub>2</sub>

No.	Tanggal Sampel	$\rho_{\text{CH}_4}$ (kg/m <sup>3</sup> )	$\rho_{\text{CO}_2}$ (kg/m <sup>3</sup> )	$C_{\text{CH}_4}$ (%)
1.	28 April 2022			8,79
2.	3 Mei 2022			11,79
3.	8 Mei 2022	0,668	1,842	44,07
4.	13 Mei 2022			50,16
5.	18 Mei 2022			54,16

Rumus (Tabatabaei, 2018) :

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

- 28 April 2022

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

$$\rho_{\text{biogas}} = 0,0879 \times 0,668 \text{ kg/m}^3 + (1 - 0,0879) \times 1,842 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 0,0587172 \text{ kg/m}^3 + 1,6800882 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 1,7388054 \text{ kg/m}^3$$

- 3 Mei 2022

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

$$\rho_{\text{biogas}} = 0,1179 \times 0,668 \text{ kg/m}^3 + (1 - 0,1179) \times 1,842 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 0,0787572 \text{ kg/m}^3 + 1,6248282 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 1,7035854 \text{ kg/m}^3$$

- 8 Mei 2022

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

$$\rho_{\text{biogas}} = 0,4407 \times 0,668 \text{ kg/m}^3 + (1 - 0,4407) \times 1,842 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 0,2943876 \text{ kg/m}^3 + 1,0302306 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 1,3246182 \text{ kg/m}^3$$

- 13 Mei 2022

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

$$\rho_{\text{biogas}} = 0,5016 \times 0,668 \text{ kg/m}^3 + (1 - 0,5016) \times 1,842 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 0,3350688 \text{ kg/m}^3 + 0,9180528 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 1,2531216 \text{ kg/m}^3$$

- 18 Mei 2022

$$\rho_{\text{biogas}} = C_{\text{CH}_4} \times \rho_{\text{CH}_4} + (1 - C_{\text{CH}_4}) \times \rho_{\text{CO}_2}$$

$$\rho_{\text{biogas}} = 0,5416 \times 0,668 \text{ kg/m}^3 + (1 - 0,5416) \times 1,842 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 0,3617888 \text{ kg/m}^3 + 0,8443728 \text{ kg/m}^3$$

$$\rho_{\text{biogas}} = 1,2061616 \text{ kg/m}^3$$

Tabel B.6 Nilai Massa Jenis Biogas

No.	Tanggal Sampel	$\rho_{\text{biogas}}$ (kg/m <sup>3</sup> )
1.	28 April 2022	1,7388054
2.	3 Mei 2022	1,7035854
3.	8 Mei 2022	1,3246182
4.	13 Mei 2022	1,2531216
5.	18 Mei 2022	1,2061616



Tabel B 7. Densitas Gas

Gas	Formula	Molecular weight	Density - $\rho$ -	
			( $\text{kg/m}^3$ )	( $\text{lb/ft}^3$ )
Acetylene (ethyne)	$C_2H_2$	26	1.092 <sup>1)</sup> 1.170 <sup>2)</sup>	0.0682 <sup>1)</sup> 0.0729 <sup>2)</sup>
Air		29	1.205 <sup>1)</sup> 1.293 <sup>2)</sup>	0.0752 <sup>1)</sup> 0.0806 <sup>2)</sup>
Ammonia	$NH_3$	17.031	0.717 <sup>1)</sup> 0.769 <sup>2)</sup>	0.0448 <sup>1)</sup> 0.0480 <sup>2)</sup>
Argon	$Ar$	39.948	1.661 <sup>1)</sup> 1.7837 <sup>2)</sup>	0.1037 <sup>1)</sup> 0.111353 <sup>2)</sup>
Benzene	$C_6H_6$	78.11	3.486	0.20643
Blast furnace gas			1.250 <sup>2)</sup>	0.0780 <sup>2)</sup>
Butane	$C_4H_{10}$	58.1	2.489 <sup>1)</sup> 2.5 <sup>2)</sup>	0.1554 <sup>1)</sup> 0.156 <sup>2)</sup>
Butylene (Butene)	$C_4H_8$	56.11	2.504	0.148 <sup>2)</sup>
Carbon dioxide	$CO_2$	44.01	1.842 <sup>1)</sup> 1.977 <sup>2)</sup>	0.1150 <sup>1)</sup> 0.1234 <sup>2)</sup>
Carbon disulphide		76.13		
Carbon monoxide	$CO$	28.01	1.165 <sup>1)</sup> 1.250 <sup>2)</sup>	0.0727 <sup>1)</sup> 0.0780 <sup>2)</sup>
Carbureted Water Gas				0.048
Chlorine	$Cl_2$	70.906	2.994 <sup>1)</sup>	0.1869 <sup>1)</sup>
Coal gas			0.58 <sup>2)</sup>	
Coke Oven Gas				0.034 <sup>2)</sup>
Combustion products			1.11 <sup>2)</sup>	0.069 <sup>2)</sup>
Cyclohexane		84.16		
Digester Gas (Sewage or Biogas)				0.062
Ethane	$C_2H_6$	30.07	1.264 <sup>1)</sup>	0.0789 <sup>1)</sup>
Ethyl Alcohol		46.07		
Ethyl Chloride		64.52		
Ethylene	$C_2H_4$	28.03	1.260 <sup>2)</sup>	0.0786 <sup>2)</sup>
Helium	$He$	4.02	0.1664 <sup>1)</sup> 0.1785 <sup>2)</sup>	0.01039 <sup>1)</sup> 0.011143 <sup>2)</sup>
N-Heptane		100.20		
Hexane		86.17		
Hydrogen	$H_2$	2.016	0.0899 <sup>2)</sup>	0.0056 <sup>2)</sup>
Hydrochloric Acid		36.47	1.63 <sup>2)</sup>	
Hydrogen Chloride	$HCl$	36.5	1.528 <sup>1)</sup>	0.0954 <sup>1)</sup>
Hydrogen Sulfide	$H_2S$	34.076	1.434 <sup>1)</sup>	0.0895 <sup>1)</sup>
Krypton			3.74 <sup>2)</sup>	
Methane	$CH_4$	16.043	0.668 <sup>1)</sup> 0.717 <sup>2)</sup>	0.0417 <sup>1)</sup> 0.0447 <sup>2)</sup>
Methyl Alcohol		32.04		

Methyl Butane		72.15		
Methyl Chloride		50.49		
Natural gas		19.5	0.7 - 0.9 <sup>2)</sup>	0.044 - 0.056 <sup>2)</sup>
Neon	Ne	20.179	0.8999 <sup>2)</sup>	0.056179 <sup>2)</sup>
Nitric oxide	NO	30.0	1.249 <sup>1)</sup>	0.0780 <sup>1)</sup>
Nitrogen	N <sub>2</sub>	28.02	1.165 <sup>1)</sup> 1.2506 <sup>2)</sup>	0.0727 <sup>1)</sup> 0.078072 <sup>2)</sup>
Nitrogen Dioxide	NO <sub>2</sub>	46.006		
N-Octane		114.22		
Nitrous Oxide	N <sub>2</sub> O	44.013		0.114 <sup>1)</sup>
Nitrous Trioxide	NO <sub>3</sub>	62.005		
Oxygen	O <sub>2</sub>	32	1.331 <sup>1)</sup> 1.4290 <sup>2)</sup>	0.0831 <sup>1)</sup> 0.089210 <sup>2)</sup>
Ozone	O <sub>3</sub>	48.0	2.14 <sup>2)</sup>	0.125
N-Pentane		72.15		
Iso-Pentane		72.15		
Propane	C <sub>3</sub> H <sub>8</sub>	44.09	1.882 <sup>1)</sup>	0.1175 <sup>1)</sup>
Propene (propylene)	C <sub>3</sub> H <sub>6</sub>	42.1	1.748 <sup>1)</sup>	0.1091 <sup>1)</sup>
R-11		137.37		
R-12		120.92		
R-22		86.48		
R-114		170.93		
R-123		152.93		
R-134a		102.03		
Sasol				0.032
Sulfur	S	32.06		0.135
Sulfur Dioxide	SO <sub>2</sub>	64.06	2.279 <sup>1)</sup> 2.926 <sup>2)</sup>	0.1703 <sup>1)</sup> 0.1828 <sup>2)</sup>
Sulfur Trioxide	SO <sub>3</sub>	80.062		
Sulfuric Oxide	SO	48.063		
Toluene	C <sub>7</sub> H <sub>8</sub>	92.141	4.111	0.2435
Water Vapor, steam	H <sub>2</sub> O	18.016	0.804	0.048
Water gas (bituminous)				0.054
Xenon			5.86 <sup>2)</sup>	

Sumber : *Engineering ToolBox*, 2012

## LAMPIRAN C DOKUMENTASI

### 1. Pembuatan Biogas

- Persiapan alat dan bahan pembuatan biogas menggunakan biodigester



- Pengambilan biogas dan *slurry* pada biodigester



- Analisa biogas menggunakan alat *Biogas Analyzer*



- Mengukur pH slurry



## 2. Analisa COD Pada *Slurry*

- Menyiapkan alat-alat yang digunakan



- Melakukan pengenceran 500 ml *slurry*



- Menimbang dan membuat larutan  $K_2Cr_2O_7$  dan FAS



- Menyiapkan sampel dan blanko untuk kemudian direfluks



- Melakukan titrasi dengan larutan FAS



- Perubahan warna sampel sebelum dititrasi dan setelah dititrasi

