

### The 3<sup>rd</sup> FIRST 2019 INTERNATIONAL CONFERENCE FORUM IN RESEARCH, SCIENCE, AND TECHNOLOGY OCTOBER 09-10, 2019

Palembang, Province of South Sumatera Indonesia



#### **PREFACE**

The 3rd FIRST 2019 (Forum in Research, Science, and Technology) International Conference was initiated and organized by State Polytechnic of Sriwijaya in collaboration with Management and Science, University (MSU), Malaysia and National Chin-Yi University of Technology, Taiwan. The theme of the conference was "Integration of Advanced Technology to Enhance Social Welfare". The 3rd FIRST 2019 International Conference facilitated the participants from all over the world to meet face to face to open chances in establishing connection and collaboration among them. It was not only for the researchers in academics, but also in industries and governments. This conference became an effective media to link the researchers from many parts of the world conference, for exchanging, sharing, following up and discussing the results of the latest research, industry's needs, and government regulatory policies. The 3rd FIRST 2019 International Conference became worthwhile platform for researchers to present their finding in the areas on multidisciplinary of Engineering and Science (Track 1), Computer Science and ICT (Track 2), and Social Science (Track 3). It has also provided an opportunity for the professionals and researchers to learn and share about the latest development and research in those 3 tracks.

The 3rd FIRST 2019 International Conference attracted so many authors not only from Indonesia but also from other countries, such as Japan, Taiwan, and Malaysia. There were 180 papers were accepted in the 3rd FIRST 2019 International Conference, including 89 papers for Track 1 (Engineering and Science), 46 papers for Track 2 (Computer Science and ICT), and 45 papers for Track 3 (Social Science). In The 3rd FIRST 2019 International Conference, there were 4 keynote speakers and 2 invited speakers. As the keynote speakers, there were Prof. Yasushi Kiyoki, Ph. D from KEIO University, Japan, Prof. Nurul Taufiqu Rochman, M. Eng, Ph. D, From LIPI, Indonesia, Prof. Tjiptohadi Sawarjuwono, M. Ec., Ph. D., Ak, from Universitas Airlangga, Indonesia, and Prof. Win-Jet Luo, from National Chin-Yi University of Technology, Taiwan. As the invited speakers, there were Dr. R. Wisnu Nurcahyo, DVM from Universitas Gadjah Mada, Indonesia, and Assoc. Prof. Dr. Intan Zaurah binti Mat Darus from Universiti Teknologi, Malaysia. The 3rd FIRST 2019 International Conference committee would like to say thank you very much for all the participants and their respected institutions that have supported for the success of the 3rd FIRST 2019 International Conference, and also forall of the guess and sponsors of the 3rd FIRST 2019 International Conference.

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### Converting Waste Cooking Oil into Biodiesel using Microwaves and High Voltage Technology

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**Abstract.** Increasing fuel demand has an impact on decreasing fossil energy reserves. One of the solution to solve this problem is by increasing the production of biomass fuels, for example, biodiesel. Biodiesel can be produced from waste cooking oil through transesterification stage that reacts oil molecules with alcohol and catalyst to produce methyl esters. Heating method that can be used is combined of microwaves and high voltage technology. Microwaves technology utilizes wave emission which is absorbed and reflected by the sample to make sample's temperature being higher than surface's temperature of reactor's wall. In the process of producing biodiesel, reaction temperature is one of the several factors that is affected for heating process. From this research, we had optimum reaction temperature to get good quality product is at 60°C with 88.88% yield, 0.88 gr/ml for density, 2.52 cSt for viscosity, 102.5°C for flash point, 0.27% for water content, and 0.33 mg-KOH/gr-sample.

Keyword: Biodiesel, microwaves, high voltage, transesterification, waste cooking oil, quality

#### 1. Introduction

Increasing fuel oil can reduce fossil energy reserves. For solving this problem, the government utilizes domestic energy by increasing biomass fuel production. Utilization of biomass material as diesel fuel (biodiesel) is one of the solutions that can be applied. Biodiesel is one of an alternative fuels that can be obtained from plant and animal fats [1]. Biodiesel is a monoalkyl ester from fatty acids that is contained in vegetable oils or animal fats for using as diesel engine fuel. Biomass resources in Indonesia is amount to 32,654 MW while installed capacity currently is only around 92,726 MW [2].

One of biodiesel raw materials that often used is waste cooking oil. Total amount of waste cooking oil from various sectors in Indonesia is 3.88 million tons per year [3]. Waste cooking oil can be converted into biodiesel through transesterification that reacts oil with alcohol and catalyst. For converting it into biodiesel, we need a method that can convert into high yield and has accordance with Indonesian National Standards (SNI). The main factors that can affect the number of conversions in transesterification are ratio between triglycerides and alcohol, type of catalyst, reaction temperature, water and free fatty acids content, and speed of stirring [4]. Alcohol must be in excess condition from its stoichiometric needed in order to get a reaction that shifts towards the product. Activation energy

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can be reduced using catalyst that we used so the reaction will run easily. The temperature during transesterification can be carried out in temperature range of 30-65°C. Temperature will make molecule's movement be faster. The amount of water content in raw material can make saponification that allows to reduce catalyst efficiency level while higher value of free fatty acid will require a lot of base catalysts. High stirring speed makes increasing the number of molecule collisions and reaction speed.

Method that commonly used to convert oil into biodiesel is conventional method. Heating process with this method is very slow because energy transfer to the material depend on convection current and thermal conductivity of reaction mixture [5]. Various alternative methods have been developed to produce biodiesel with high yield in a short time. Hartono, et al. in 2013 develop biodiesel synthesis using an interesterification method to optimize the formation of biodiesel yield [6]. However, this method is not efficient because the time to process biodiesel is quite long, which is 60 minutes for 87.63% yield. In 2015, Kalsum et al. using a reactor membrane to produce biodiesel from waste cooking oil [7]. However, this study is not efficient because it takes 2.5 hours to produce biodiesel of 94.81% yield. Moeksin, et al. (2017) has done research on biodiesel production process using electrolysis method. However, the highest yield produced using this method is only 38.3% at 12 V electricity voltage [8]. Another method that has been applied is using the Continuous Microwave Biodiesel Reactor (CMBR) [9]. This method can produce biodiesel in a shorter time. However, the highest yield of triglycerides is still relatively low, which is 65.5%. So, the authors done a research of producing biodiesel using a combination of microwaves and high voltage.

Microwaves utilizes heating process that occurs directly without contact to the reactor's wall so the reaction can be mixed homogenly. This makes time needed to carry out the process significantly reduced [10]. Microwave utilization can make reaction be more efficient with reaction time and separated process in a shorter time, reduce the number of side product, and energy consumption [11]. The separated process of biodiesel with glycerol can be carried out using high voltage. High voltage utilization can make the saturation of glycerine being faster [12]. Previous research shows that 9 kV has a percentage 98% in 25 seconds [13]. In this research, six experiments were done by reaction temperature variation to get a high yield and quality that is in accordance with SNI.

### 2. Methods and Materials

The procedure for producing biodiesel from waste cooking oil using microwaves and high voltage utilization is presented in Figure 1.

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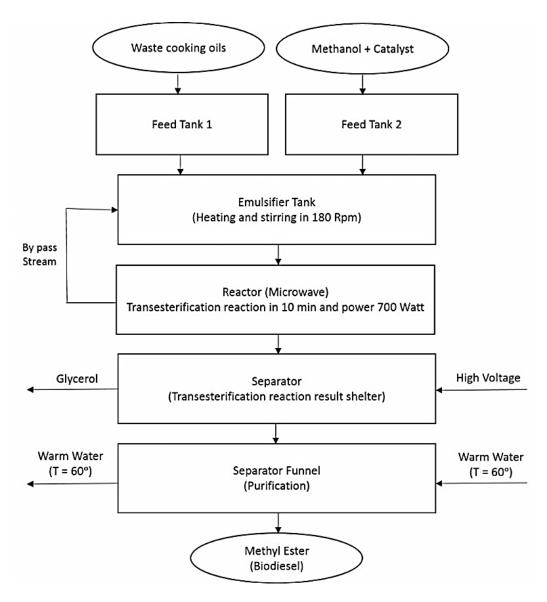


Figure 1. Biodiesel Production Process Diagram

Raw material that used is waste cooking oil from crackers factory in Kenten, Palembang, Indonesia. Catalyst as a support material in this process is sodium methoxide from reaction of 1% sodium hydroxide and methanol from weight basis of oil. Emulsifier tank has function as a place to pre-heat materials that we used. The tank is also equipped with a propeller so the result will be a homogeneous mixture between oil, methanol, and catalyst. In this process, transesterification occurs in a microwave reactor. The power used is 700 watts with reaction time of 10 minutes. In this section, the mixture will absorb and reflect wave radiation that emitted by microwave. Separator as a place of separation utilizes high voltage from cathode and anode plate that is located on the right and left sides of the tank. The number of each electrode is 3 pieces with 1 cm of distance between anode - cathode and voltage used is 10 kV. Instrumentation that used in producing biodiesel with microwaves and high voltage utilization is shown in Figure 2.

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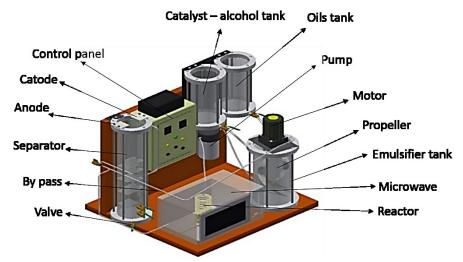


Figure 2. Biodiesel Production using Microwaves and High Voltage Instrumentation

### 2.1 Product Quality Analysist

### 2.1.1 Density Determination

Density is one of the important parameters that must be known from an oil fuel product. The density that allowed for biodiesel according to SNI is 850-890 kg/m³ at 40°C. According to Hadrah, et al. (2018), the value of density that in accordance with SNI can produce a perfect combustion reaction on engine while density value that exceeds the standard will cause the combustion reaction to be imperfect so it will increase emissions and wear on the engine [14]. In 2018, Mustariani, et al. has done research in biodiesel production from waste cooking oil using conventional methods. Heating process that carried out at 55°C will produce biodiesel with density value that still in permitted range, which is 880 kg/cm³.

### 2.1.2 Viscosity Determination

Viscosity determines thickness level of an oil fuel. Viscosity can be determined using falling ball method. Based on SNI, range that allowed for biodiesel viscosity value is 2.3-6.0 cSt at 40°C. High viscosity value will make it difficult to flow, pumping process, and fuel ignition while if it is too runny, fuel will be difficult to burn and can cause leaks in the injection pipe [15]. According to research that done by Kartika and Widyaningsih in 2012 that using 60°C as reaction temperature, the value of viscosity was 3.06 cSt and still in the range of values that is permitted by SNI [16].

### 2.1.3 Flashpoint Determination

Biodiesel flash point was determined by using Pensky-Martens closed cup instrument. Flash point need to be known because it will affect storage process of biodiesel fuel. Flash point that is lowest will make storage process of biodiesel fuel being more difficult because it will be flammable while highest flash point will complicate fuel during the ignition process. In SNI, flash point value has been set with minimum value 100°C. Research by Zulhardi, et al. shows that in the temperature range of 60-70°C, flashpoint of biodiesel will be at 227°C [17].

#### 2.1.4 Water Content Determination

Water content in biodiesel will affect the formation of soap as an unwanted product in process of biodiesel production. High water content will allow hydrolysis of triglycerides become free fatty acids and glycerol [17]. Water content that is permitted by SNI is not more than 0.05%. Based on the journal Chhetri, et al., by using 60°C as reaction temperature will produce biodiesel with 0% water content so it is still in permissible range [18].

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#### 2.1.5 Acid Number Determination

Low acid number indicates content of free fatty acids that contained in biodiesel fuel has decreased. High acid number can complicate the process of separation between biodiesel and glycerol so net result obtained will decrease [14]. The number of acids that permitted by SNI is not more than 0.5 mgKOH/gr. Meng, et al., research in 2008 showed that the reaction using temperature 50°C for 90 minutes produced biodiesel with an acid number of 0.48 mg KOH/gr [19].

### 3. Result and Discussion

### 3.1 Effect of Reaction Temperature for Biodiesel Yield

Reaction temperature will affect transesterification that is occured. Reaction temperature that used in this study are 35, 40, 45, 50, 55, and 60°C. Relationship between reaction temperature and percent of biodiesel yield can be seen in Figure 3.

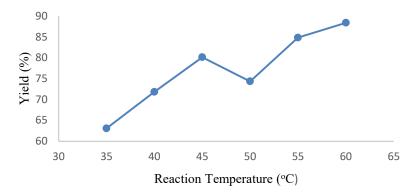


Figure 3. Effect of Reaction Temperature for Biodiesel Yield

Based on Figure 3, increasing reaction temperature will cause increasing in percent yield of biodiesel. High percentage of yield is due to faster movement of reactant molecules. This is in accordance with Wahyuni et al. (2015) that higher temperature causes faster molecular movement or kinetic energy by reacting molecules increased, so collisions between reacting molecules also being increased [20]. The collision between these reacting molecules will produce heat which is used to convert oil into biodiesel. Increasing temperature will increase the rate of reaction that occurs due to an increasing in kinetic energy system [21]. The lowest yield percentage at 50°C is due to imperfect transesterification which results in low conversion due to a shift in reaction towards saponification which produces more soap than biodiesel.

### 3.2 Effect of Reaction Temperature for Biodiesel Density

Density analysis has a function to find out how combustion reaction that occurs in diesel engine combustion chamber. High density value will cause combustion reaction to be imperfect so it can increase engine emissions and wear [14]. Higher conversion was gained when biodiesel density decreased into lower value [22]. Effect of reaction temperature on biodiesel density is shown in Figure 4.

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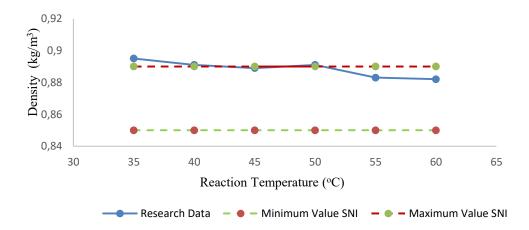


Figure 4. Effect of Reaction Temperture for Biodiesel Density

Figure 4 shows higher reaction temperature will decrease biodiesel product density. Decreasing biodiesel density is caused by faster movement of molecular substances due to an increasing temperature which causes the molecules to collide with each other [23]. This movement makes molecules of the substance stretch and decreases density of the substance. Lowest density is in biodiesel with a reaction temperature of 60°C.

### 3.3 Effect of Reaction Temperature for Biodiesel Viscosity

Viscosity will affect fuel injection process on the engine. High viscosity will hamper vehicle pump system and make engine hard to turn on [12]. Based on analysis of product characteristics, reaction temperature affects viscosity of biodiesel. Effect of reaction temperature on biodiesel viscosity can be seen in Figure 5.

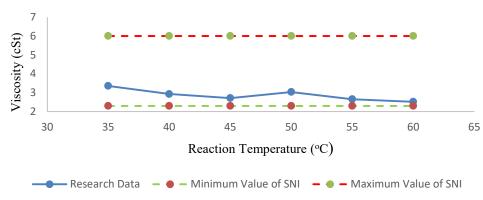


Figure 5. Effect of Rection Temperature for Biodiesel Viscosity

Figure 5 shows that higher reaction temperature will get lower viscosity of biodiesel product. Decreasing viscosity value is caused by faster movement of substance molecules due to an increasing temperature. This faster movement will increase pressure so the molecules will expand and wide the distance between molecules [24].

#### 3.4 Effect of Reaction Temperature for Biodiesel Flash Point

Flash point will affect the treatment of biodiesel product storage. Lowest flash point will complicate storage process because of its flammability, but when its too high, it will make difficulty for engine to ignite the fuel [13].

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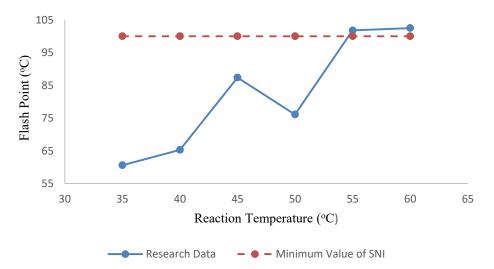


Figure 6. Effect of Reaction Temperature for Biodiesel's Flash Point

Based on the result of product characteristics analysist, it is known that reaction temperature affects biodiesel flash point. Analysis result shows that higher reaction temperature was used, flash point will be lower. High reaction temperature will make converted triglyceride molecules into methyl esters being higher in shorter time [25].

In Figure 6, biodiesel products with reaction temperature 55 and 60°C are in accordance with SNI standards. Biodiesel products with reaction temperatures 55 and 60°C have higher flash point due to the process of converting triglycerides and methanol to methyl esters being perfect enough, so methanol which does not react with triglycerides, residual catalysts, and pollutants being less. However, flash point of biodiesel products with a reaction temperature of 35, 40, 45, and 50°C are not in accordance with SNI. This value shows that impurity compounds contained in biodiesel products such as methanol which does not react is still quite large at these temperatures. This is because reaction is not perfect, so the conversion of triglycerides to methyl esters being low.

### 3.5 Effect of Reaction Temperature for Biodiesel Water Content

Based on Figure 7, it is known that reaction temperature affects water content in biodiesel. Higher reaction temperature was used, lower water content will be get. Water content that is contained in biodiesel products is caused by presence of side product from production of sodium methoxide catalyst. This water content will affect and inhibit the reaction because water will hydrolyze methyl esters that were produced [14]. If water content is lower, possibility of a saponification that forms soap due to reaction with catalyst, waste cooking oil, and residual water contained has been minimal [26]. The highest water content is found in biodiesel with reaction temperature 50°C which is 0.4335 while the lowest is found in 60°C which is 0.265%. High water content is caused by shift of the reaction towards saponification so some of the reaction turns into soap and will reduce percentage of biodiesel yield [4].

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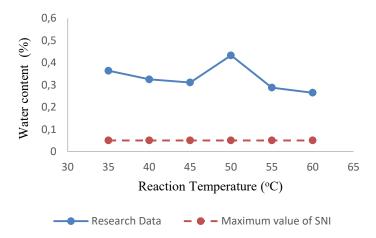


Figure 7. Effect of Reaction Temperature for Biodiesel's Water Content

Based on Figure 7 above, biodiesel products does not including SNI. All of the products still have water content that exceeds permitted limit. High water content of the product is caused by amount of water remaining in product. Existence of water can be caused by the presence of remaining water produced by purifining biodiesel. The remaining water will make machine being difficult to ignite because it will inhibit delivery of fuel to the piston.

### 3.6 Effect of Reaction Temperature for Biodiesel Acid Number

Based on the analysis of biodiesel characteristics, reaction temperature affects acid number of biodiesel. Acid number is an indication of the number of fatty acids that is contained in biodiesel products [27]. Acid number indicates that biodiesel product is not corrosive so it will not cause damage to the engine injector [28]. The higher reaction temperature was used, the lower acid number will be get. The highest acid number is in biodiesel with reaction temperature 35°C which is 0.981 mg-KOH/gr-sample while the lowest acid number is in 60°C which is 0.330 mg-KOH/gr-sample.

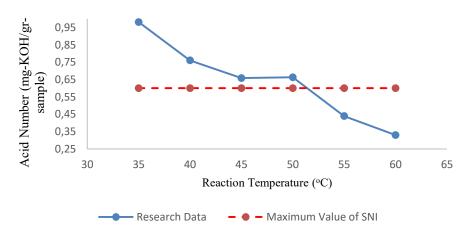


Figure 8. Effect of Reaction Temperature for Biodiesel's Acid Value

Figure 8 shows that biodiesel products with reaction temperatures 55 and 60°C are in accordance with SNI. These results indicate that the product is not corrosive to the engine injector. Meanwhile, other biodiesel products have not been in accordance with SNI. This is possible due to incompleteness of the process for converting triglyceride molecules into methyl esters so acid numbers are still quite high that is contained in product.

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#### 4. Conclusions

Utilization of biomass material as diesel fuel is one of the solutions that has potential to be applied in order to keep energy needed. Microwaves and high voltages can be combined to produce biodiesel from waste cooking oil in a short time and get high yield. Heating process using microwave method utilizes wave radiation absorbed and reflected by the mixture so contact does not occur with reactor wall. The optimum reaction temperature for getting good quality product is 60°C with 0.88 gr/ml density, 2.52 cSt viscosity, 102.5°C flash point, 0.27% water content, and 0.33 mg-KOH/gr-sample acid number. Water content has not accordance with SNI, but this can be handled by adding a refining process by heating biodiesel products so the water that contained in the product can be reduced.

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### CERTIFICATE OF APPRECIATION

Present to

### LEILA KALSUM

in recognition & appreciation of the contribution as

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